

Geo-Engineering of Climate
AOSC 433/633 & CHEM 433

Ross Salawitch

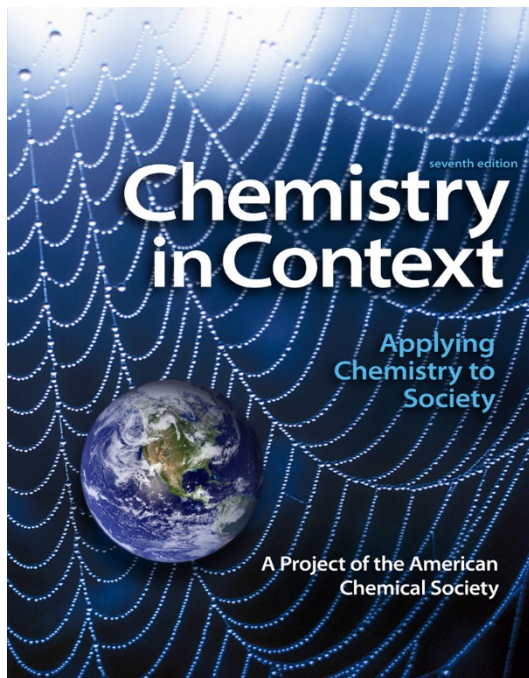
Class Web Site: <http://www.atmos.umd.edu/~rjs/class/spr2017>

Today:

- **Geo-engineering of climate**
- **Lecture designed to serve as a “mini review” of class material**

Lecture 23
9 May 2017

Course Logistics



Chemistry in Context : Applying Chemistry to Society, 7/e

American Chemical Society (ACS)

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The author team truly benefitted from the expertise of a wider community. We extend our thanks to the following individuals for the technical expertise they provided to us in preparing the manuscript:

Mark E. Anderson, University of Wisconsin--Madison

David Argentar, Sun Edge, LLC

Marion O'Leary, Carnegie Institution for Science

Ross Salawitch, University of Maryland

Kenneth A. Walz, Madison Area Technical College

- If you've rented, **please bring with you to final exam, on Wed 17 May, 10:30 am** (this room)
- Thurs lecture will be class review

Geo-engineering of weather & climate has a long history:

- 1945: John von Neumann and other leading scientists meet at Princeton and agreed that modifying weather deliberately might be possible (motivation was “next great war”)
- 1958: US Congress funded expanded rainmaking research (Irving Langmuir, GE)
- Cold War: U.S. military agencies devoted significant funds to research on what came to be called "climatological warfare"
 - one aim was to make the Arctic Ocean navigable by eliminating the ice pack
 - extensive cloud-seeding conducted over Ho Chi Minh Trail during Vietnam war, to increase rainfall and bog down the North Vietnamese Army's supply line in mud
- 1975: Mikhail Budyko calculated that if global warming ever became a serious threat, we could counter with just a few airplane flights a day in the stratosphere, burning sulfur to make aerosols that would reflect sunlight away
- 1977: N.A.S. report looked at a variety of schemes to reduce global warming, should it ever become dangerous, and concluded a turn to renewable energy was a more practical solution than geo-engineering of climate

Source: S. Weart, The Discovery of Global Warming, Harvard University Press, 2003
<http://www.aip.org/history/climate/>

Geo-engineering of weather & climate has a long history:

Stephen Schneider, Geo-engineering: could –or should – we do it ?,
Climatic Change, **33**, 291, 1996:

Although I believe it would be irresponsible to implement any large-scale geo-engineering scheme until scientific, legal, and management uncertainties are substantially narrowed, I do agree that, given the potential for large inadvertent climatic changes now being built into the earth system, more systematic study of the potential for geo-engineering is probably needed.

Geo-engineering of weather & climate has a long history:

Two general classifications:

- **Modification of surface radiative forcing as CO₂ rises**
 - **space shield blocking portion of solar irradiance**
 - **stratospheric balloons blocking portion of solar irradiance**
 - **injection of sulfate particles into stratosphere to ↑ albedo**
 - **modification of tropospheric clouds to ↑ albedo**
- **Carbon control and /or sequestration**
 - **iron fertilization of oceans**
 - **carbon burial**

Since August 2006:

- **Nov 2006: Geo-engineering workshop, NASA Ames**

- led by Robert Chatfield and Max Loewenstein
- 40 page workshop report (<http://event.arc.nasa.gov/main/home/reports/SolarRadiationCP.pdf>)

- **Oct 2007: Ken Caldeira, NY Times Op Ed**

- Seeding the stratosphere might not work perfectly ... but is cheap, easy and worth investigating...
- Think of it as an insurance policy, a backup plan for climate change.
- Which is the more environmentally sensitive thing to do: let the Greenland ice sheet collapse and polar bears become extinct, or throw a little sulfate in the stratosphere? The second option is at least worth looking into.

<http://www.nytimes.com/2007/10/24/opinion/24caldiera.html>

- **Nov 2007: Geo-engineering meeting, Harvard University**

- covered by Science (<http://sciencenow.sciencemag.org/cgi/content/full/2007/1109/1>)

Harvard climate researcher James Anderson told the group that the arctic ice was "holding on by a thread" and that more carbon emissions could tip the balance.

The delicacy of the system, he said "convinced me of the need for research into geo-engineering" And 5 years ago? "I would have said it's a very inappropriate solution"

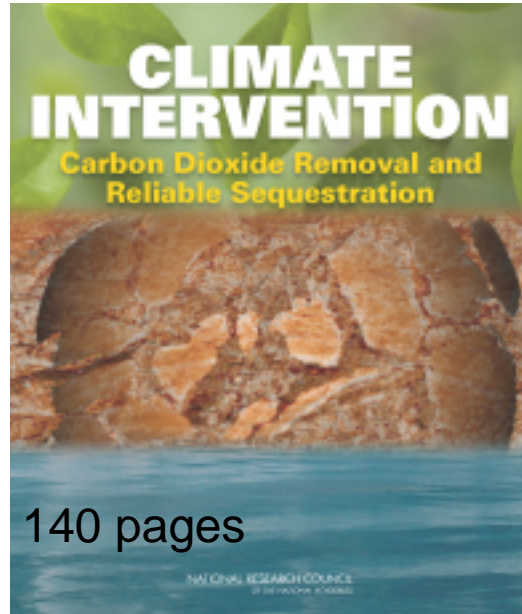
- **June 2009: National Academy of Sciences (NAS) Geo-engineering meeting**

- Chapter 15, Solar Radiation Management (SRM) of NAS America Climate Choice's 2010 report:

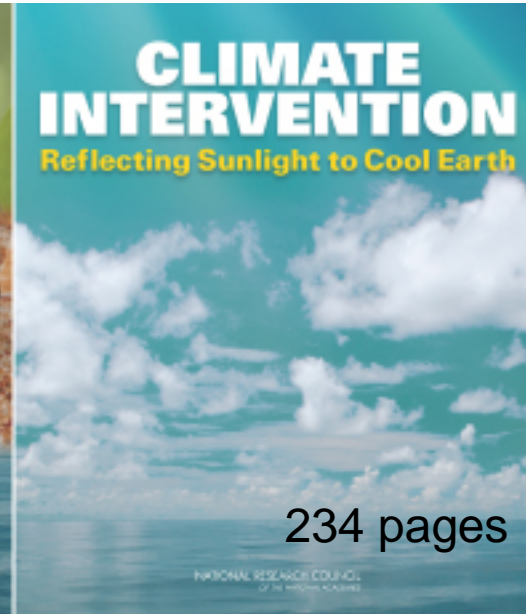
Little is currently known about the efficacy or potential unintended consequences of SRM approaches, particularly how to approach difficult ethical and governance questions. Therefore, research is needed to better understand the feasibility of different approaches; the potential consequences of such approaches on different human and environmental systems; and the related physical, ecological, technical, social, and ethical issues, including research that could inform societal debates about what would constitute a "climate emergency" and on governance systems that could facilitate whether, when, and how to intentionally intervene in the climate system.

Since August 2006:

- **Feb 2015: Two “Climate Intervention” reports issued by the prestigious National Academy of Sciences**



140 pages



234 pages

Box 2. Carbon Dioxide Removal Strategies Considered in This Study

- Changes in land use management to enhance natural carbon sinks such as forests and agricultural lands
- Accelerated weathering in the ocean and on land to enhance natural processes that remove carbon dioxide from the atmosphere
- Bioenergy with carbon capture and sequestration
- Direct air capture and sequestration of carbon dioxide
- Ocean iron fertilization to boost phytoplankton growth and enhance take-up of carbon dioxide

Box 3. Albedo Modification Strategies Considered in This Study

- Stratospheric aerosols that help reflect sunlight back into space
- Marine cloud brightening to enhance reflection of sunlight

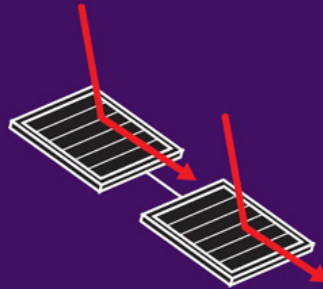
Since August 2006:

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Six recommendations:

1. Efforts to address climate change should continue to focus most heavily on mitigating GHG emissions in combination with adapting to the impacts of climate change because these approaches do not present poorly defined and poorly quantified risks and are at a greater state of technological readiness
2. Research and development investment to improve methods of CO₂ removal and disposal at scales that would have a global impact on reducing greenhouse warming, in particular to minimize energy and materials consumption, identify and quantify risks, lower costs, and develop reliable sequestration and monitoring
3. Albedo modification at scales sufficient to alter climate should not be deployed at this time
4. An albedo modification research program be developed and implemented that emphasizes multiple benefit research that also furthers both basic understanding of the climate system and its human dimensions
5. United States improve its capacity to detect and measure changes in radiative forcing and associated changes in climate
6. Initiation of a serious deliberative process to examine:
 - (a) What types of research governance, beyond those that already exist, may be needed for albedo modification research;
 - (b) The types of research that would require such governance, potentially based on the magnitude of their expected impact on radiative forcing, their potential for detrimental direct and indirect effects, and other considerations

Ways to Cool the Planet



SPACE SHIELDS

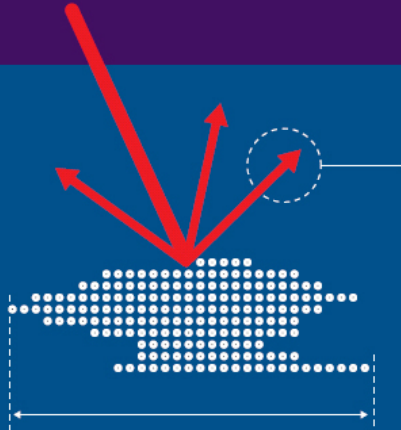
Steerable micrometers-thick refractive screens could divert a portion of the sun's energy away from Earth thus cooling the atmosphere. The screens would orbit between the sun and the Earth.

- ▲ No pollution; can be turned on or off quickly.
- ▼ Even using futuristic launching technology, the 20 million metric tons of mesh would cost US \$4 trillion to deploy.

PARTICLES IN THE STRATOSPHERE

Sulfate or other reflective particles injected at the equator stay aloft in the stratosphere for one or two years, reflecting sunlight and cooling the planet.

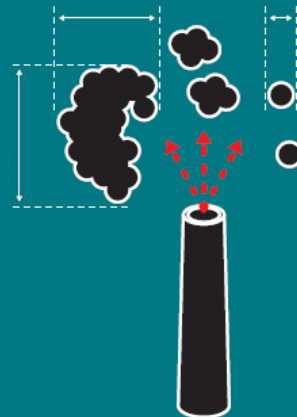
- ▲ Principle proven by volcanic eruptions; \$130 billion price tag is relatively reasonable.
- ▼ Increased acid rain, ozone layer damage.



REFLECTIVE BALLOONS

Reflective balloons would bounce a portion of the sun's energy away from Earth before it had a chance to warm the surface or the lower atmosphere.

- ▲ Cheaper to launch than space shields or space dust.
- ▼ Would require millions of balloons that would eventually fall to Earth as trash.



CLOUD COVER

Ships spray salt-water droplets that make ocean clouds more long-lasting and reflective, cooling the planet.

- ▲ Pollution free.
- ▼ Would take some 5000 salt-water spraying ships, at \$2 million to \$5 million apiece, to counter a carbon dioxide doubling.

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by Paul J. Crutzen : Climatic Change, 77, 211-219, 2006

According to model calculations ... complete *improvement in air quality* could lead to a decadal global average surface air temperature increase by 0.8 K on most continents and 4 K in the Arctic. Further studies indicate that global average climate warming during this century may even surpass the highest values in the projected IPCC global warming range of 1.4–5.8°C

**What aspect of air quality improvement
might lead to a large increase
in surface air temperature?**

Volcanic Cooling used as a Surrogate for Geo-Engineering of Climate

Albedo Enhancement by Stratospheric Sulfur Injections: A Contribution to Resolved a Policy Dilemma?

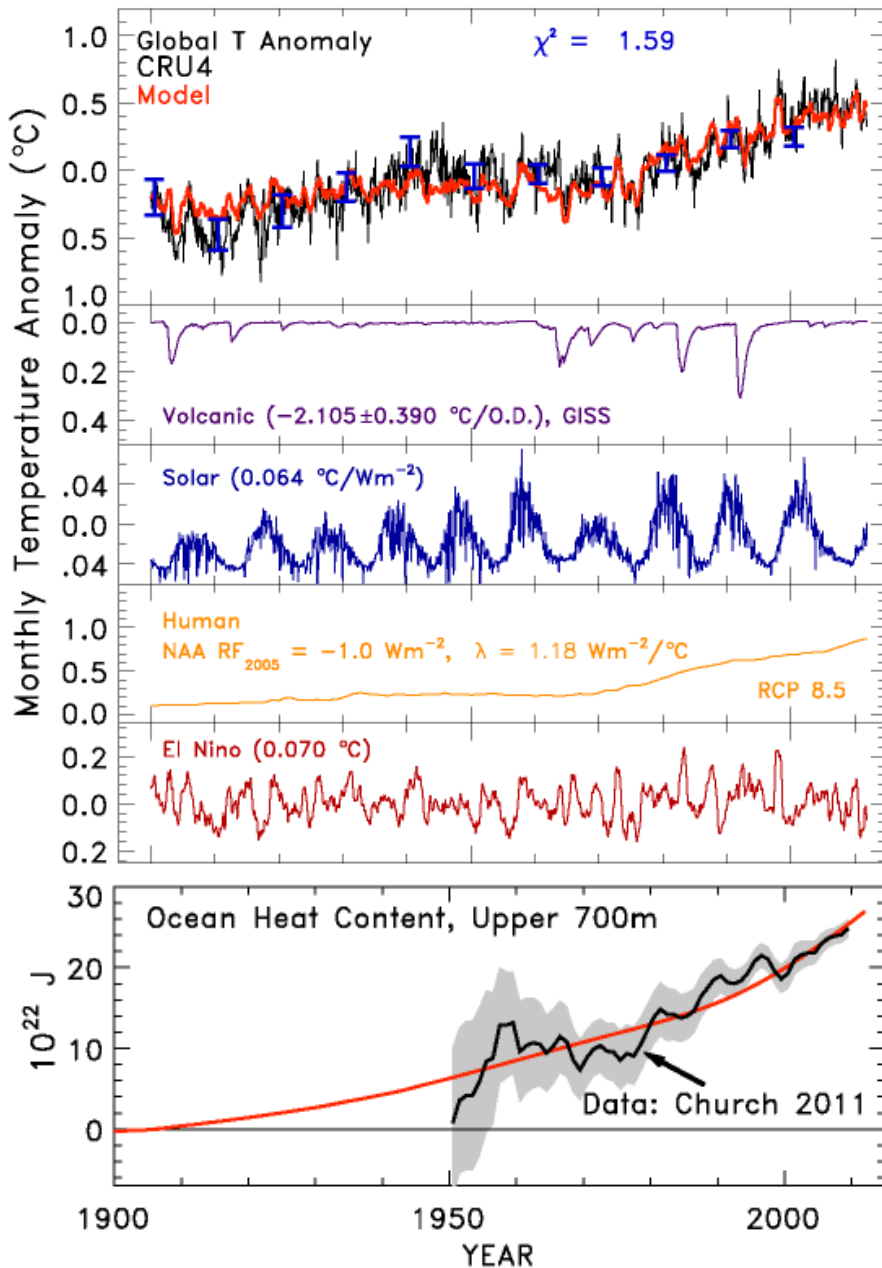
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Mount Pinatubo in June, 1991, which injected some 10 Tg S, initially as SO₂, into the tropical stratosphere (Wilson et al., 1993; Bluth et al., 1992). In this case enhanced reflection of solar radiation to space by the particles cooled the earth's surface on average by 0.5 °C in the year following the eruption (Lacis and Mishchenko, 1995).

Scientific Echo Chamber: Major Volcanic Eruptions Cause ~0.5°C Drop In Global Surface Temperature

The most dramatic change in aerosol-produced reflectivity comes when major volcanic eruptions eject material very high into the atmosphere. Rain typically clears aerosols out of the atmosphere in a week or two, but when material from a violent volcanic eruption is projected far above the highest cloud, these aerosols typically influence the climate for about a year or two before falling into the troposphere and being carried to the surface by precipitation. Major volcanic eruptions can thus cause a drop in mean global surface temperature of about half a degree celsius that can last for months or even years.

page 97, Chapter 1,
Historical Overview of Climate Change Science,
IPCC Physical Science Basis, 2007



First shown in Lecture 2.
Also shown Lectures 7 & 8

$$\Delta T_{\text{MDL } i} = (1 + \gamma) (\text{GHG RF}_i + \text{NAA RF}_i) / \lambda_{\text{BB}} + C_0 + C_1 \times \text{SOD}_{i-6} + C_2 \times \text{TSI}_{i-1} + C_3 \times \text{ENSO}_{i-2} - Q_{\text{OCEAN } i} / \lambda_{\text{BB}}$$

where

$$\lambda_{\text{BB}} = 3.21 \text{ W m}^{-2} / ^{\circ}\text{C}$$

$$1 + \gamma = \{ 1 - \Sigma(\text{Feedback Parameters}) / \lambda_{\text{BB}} \}^{-1}$$

NAA RF = net RF due to anthropogenic aerosols

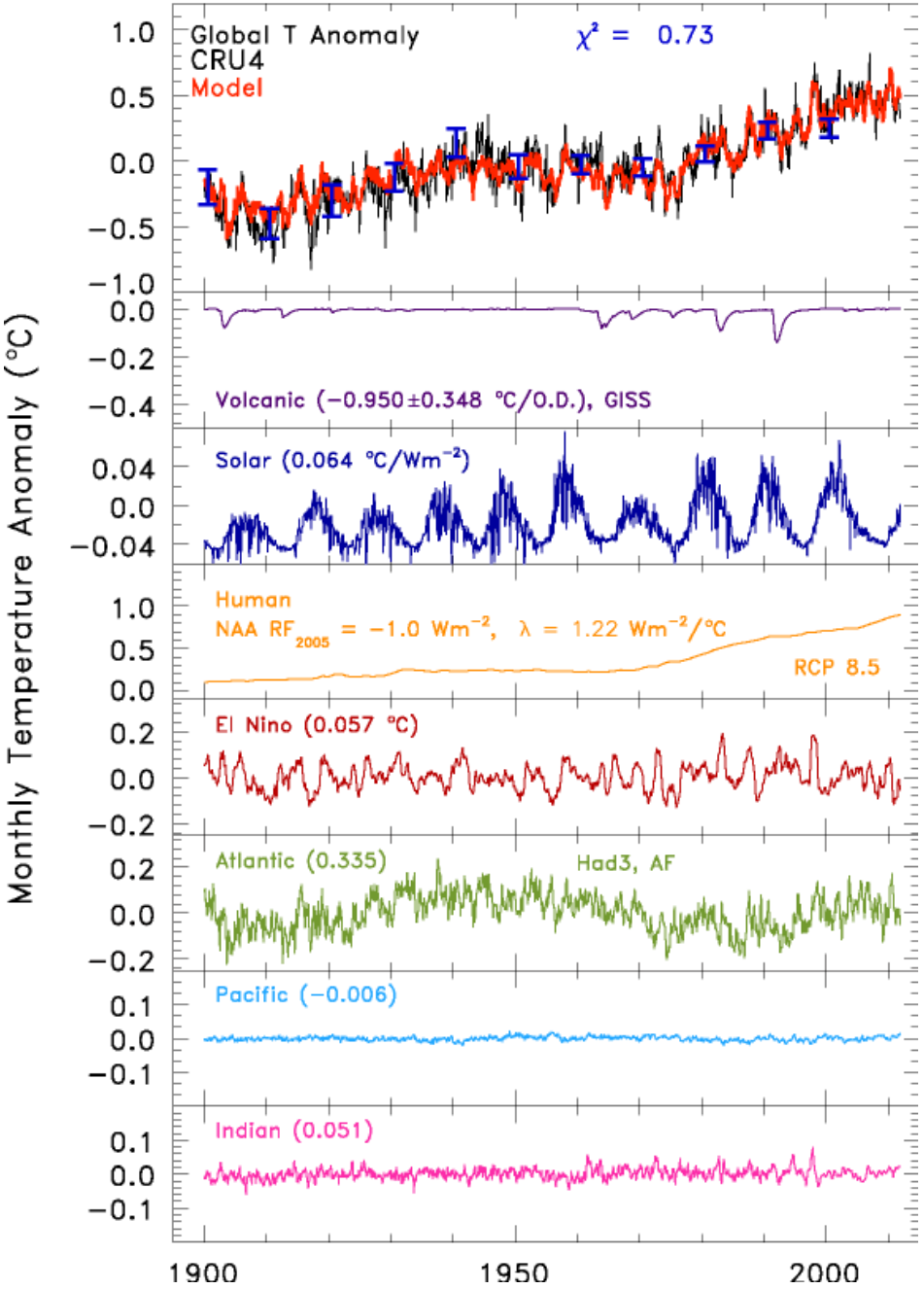
SOD = Stratospheric optical depth

TSI = Total solar irradiance

ENSO = Multivariate El Niño South. Osc Index

Q_{OCEAN} = Export of heat from atmosphere to ocean

First shown in Lecture 2.
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$$\Delta T_{MDL_i} = (1 + \gamma) (GHG RF_i + NAA RF_i) / \lambda_{BB} + C_0 + C_1 \times SOD_{i-6} + C_2 \times TSI_{i-1} + C_3 \times ENSO_{i-2} + C_4 \times AMV_i + C_5 \times PDO_i + C_6 \times IOD_i - Q_{OCEAN_i} / \lambda_{BB}$$

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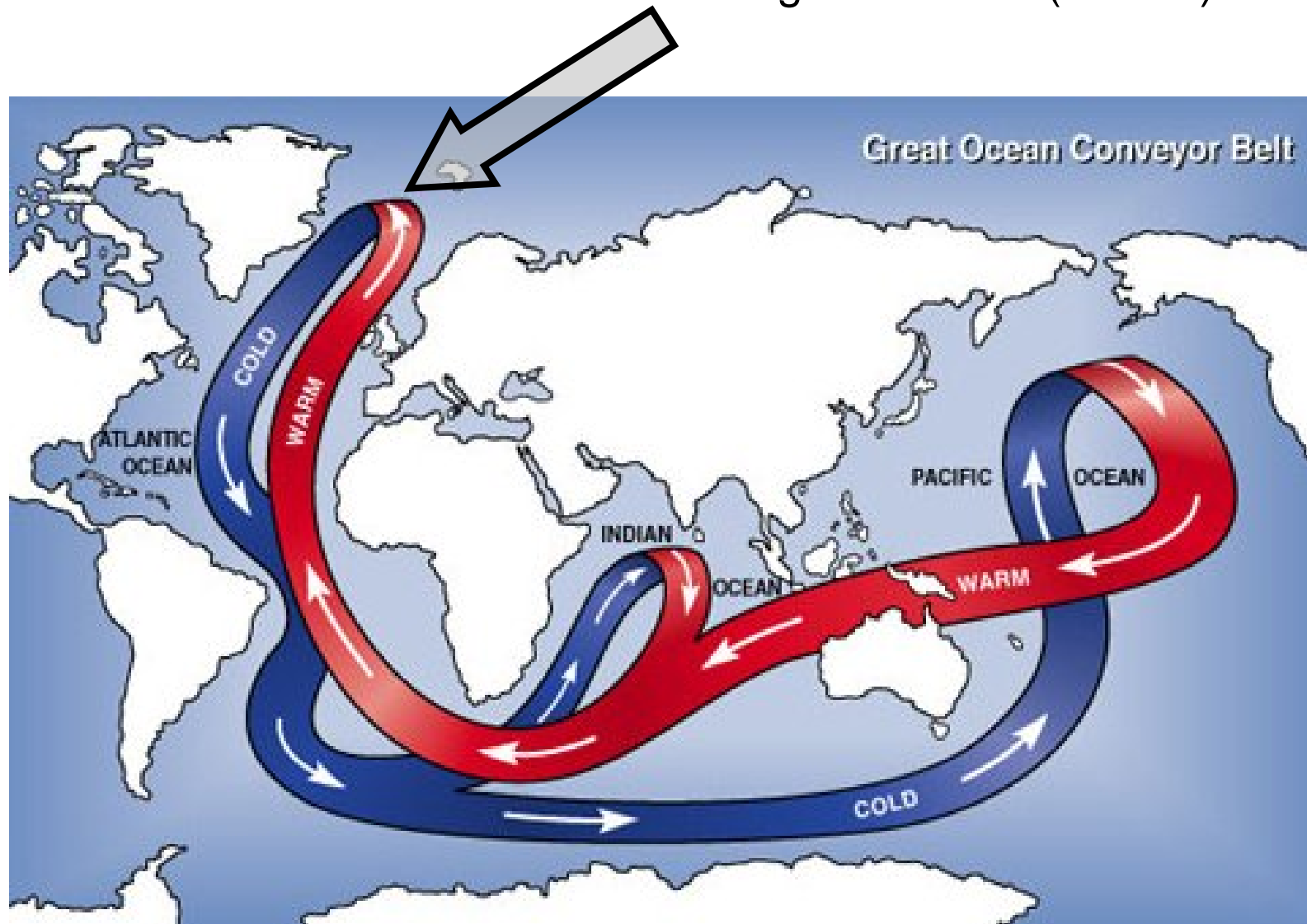
AMV = Atlantic Multidecadal Variation

PDO = Pacific Decadal Oscillation

IOD = Indian Ocean Dipole

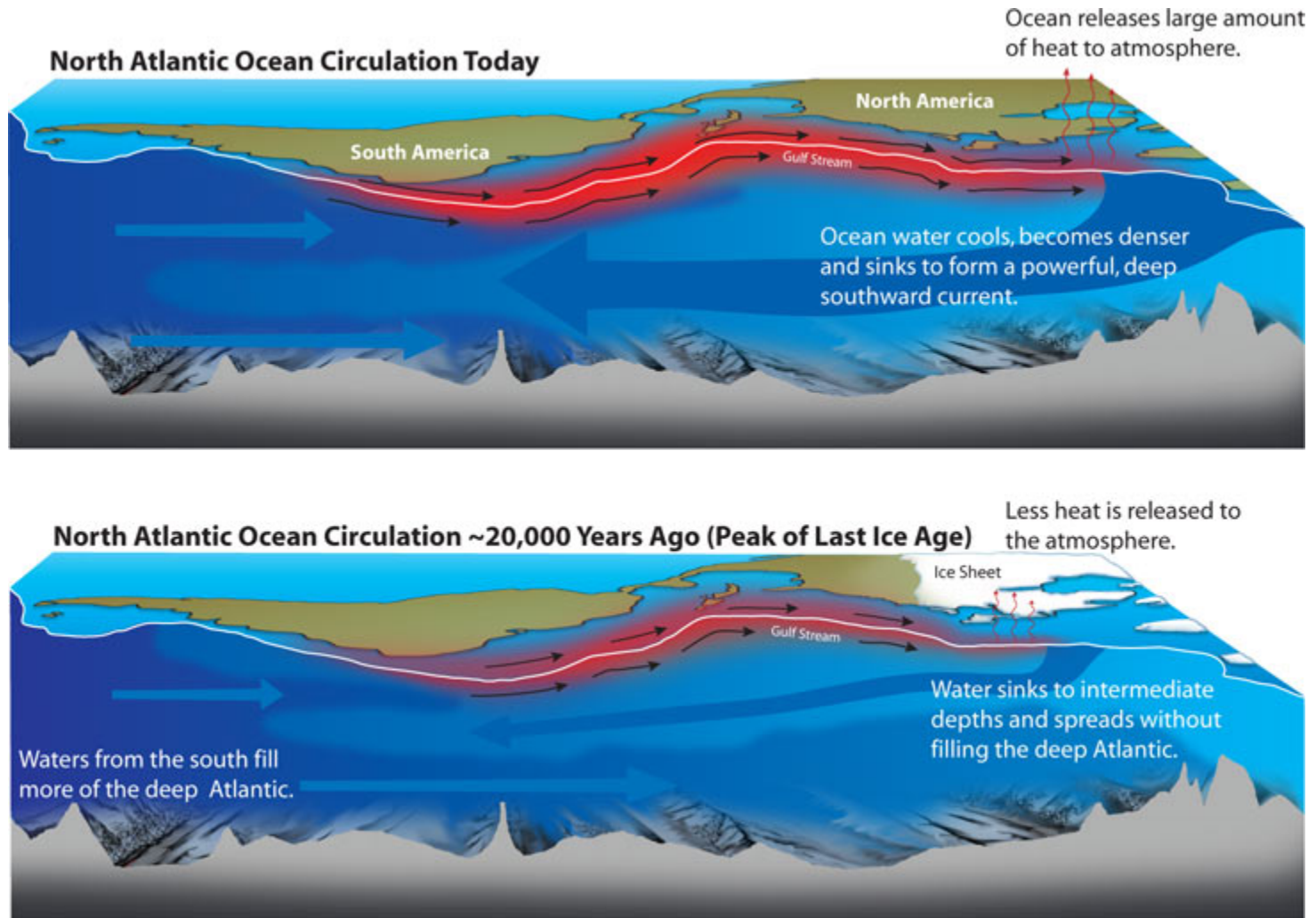
Canty et al., ACP, 2013

Atlantic Meridional Overturning Circulation (AMOC)



Lecture 5

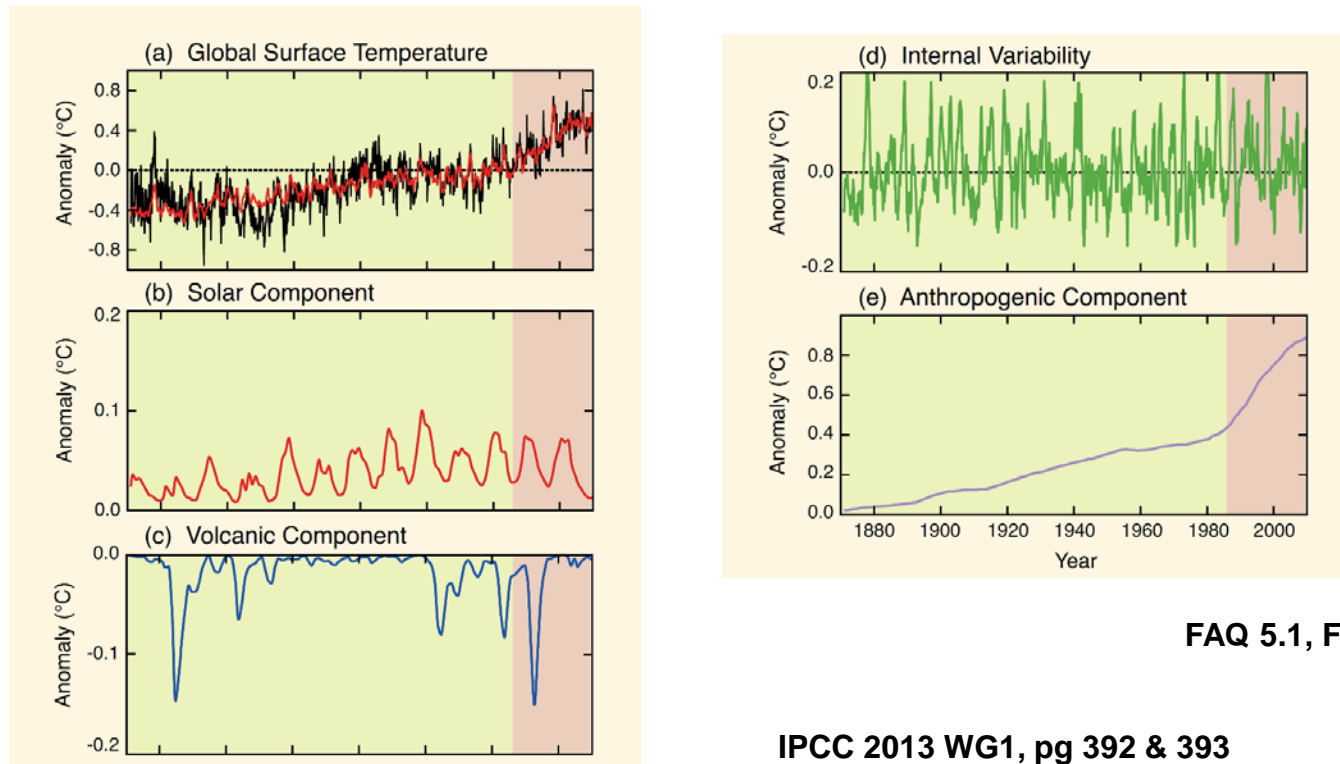
Atlantic Meridional Overturning Circulation (AMOC)



http://www.whoi.edu/cms/images/oceanus/2006/11/nao-en_33957.jpg

0.5°C cooling after Pinatubo is Science Fiction !

IPCC (2013) states Pinatubo caused global surface T to fall by 0.1 to 0.3°C, consistent with our work



FAQ 5.1, Figure 1

IPCC 2013 WG1, pg 392 & 393

Volcanic eruptions contribute to global surface temperature change by episodically injecting aerosols into the atmosphere, which cool the Earth's surface (FAQ 5.1, Figure 1c). Large volcanic eruptions, such as the eruption of Mt. Pinatubo in 1991, can cool the surface by around 0.1°C to 0.3°C for up to three years. *(continued on next page)*

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- Mt Pinatubo: $\Delta S_{\text{STRATOSPHERE}} \approx 6 \text{ Tg} \Rightarrow 4.5 \text{ W m}^{-2} \downarrow$ surface radiative forcing
0.5 °C cooling
- Doubling CO_2 will result in $\sim 3.7 \text{ W m}^{-2} \uparrow$ surface radiative forcing

$$\Delta F \approx 5.35 \text{ W m}^{-2} \ln \left(\frac{\text{CO}_2^{\text{Final}}}{\text{CO}_2^{\text{Initial}}} \right)$$

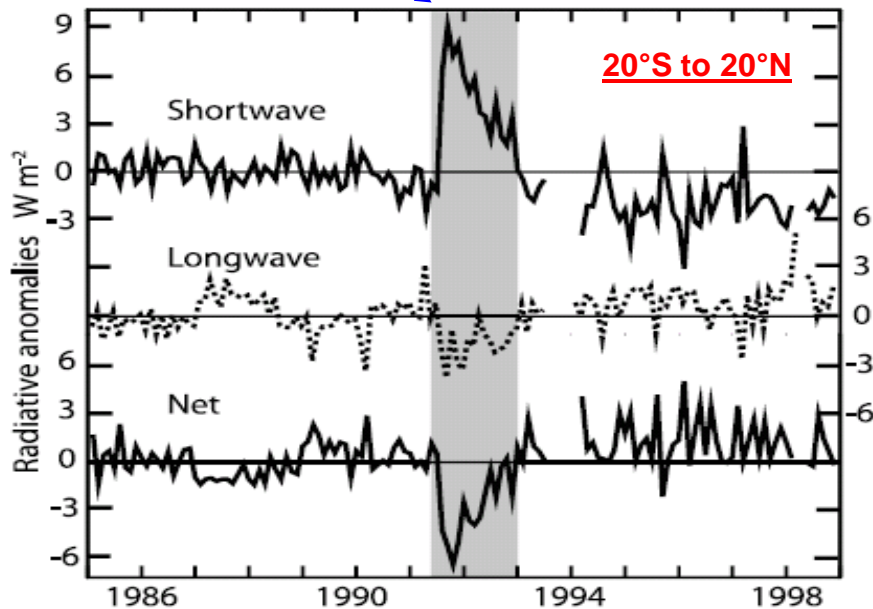
Lecture 4

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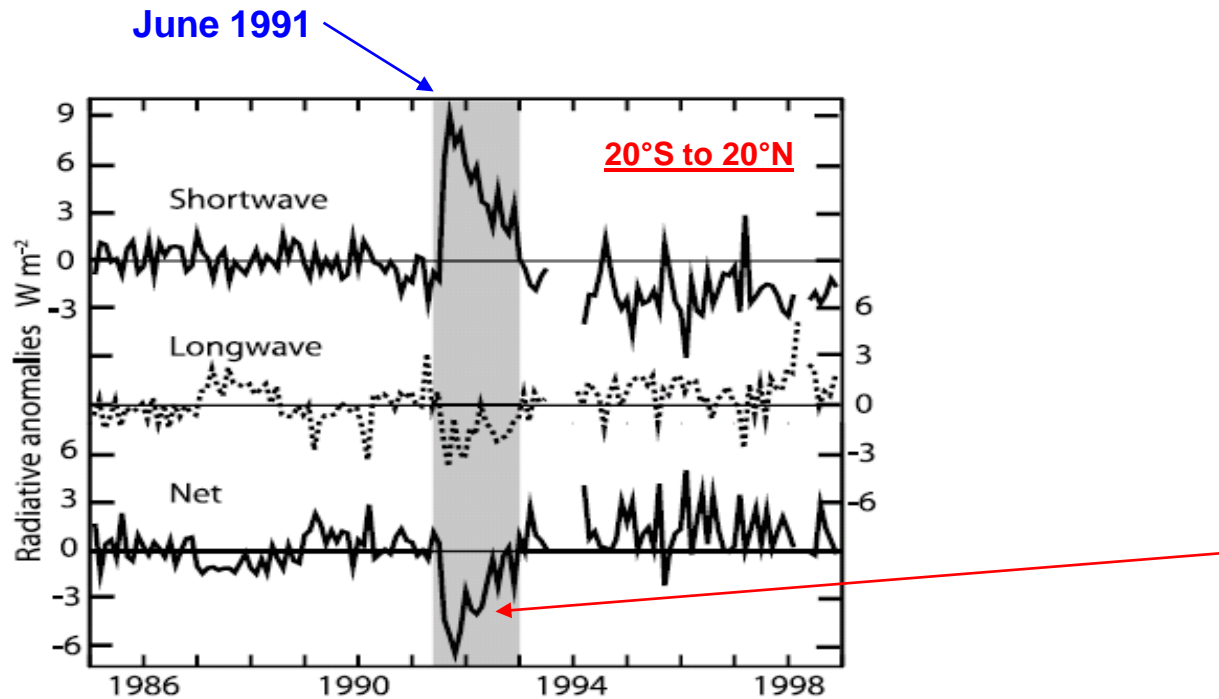
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June 1991



Trenberth and Dai, *GRL*, 2007

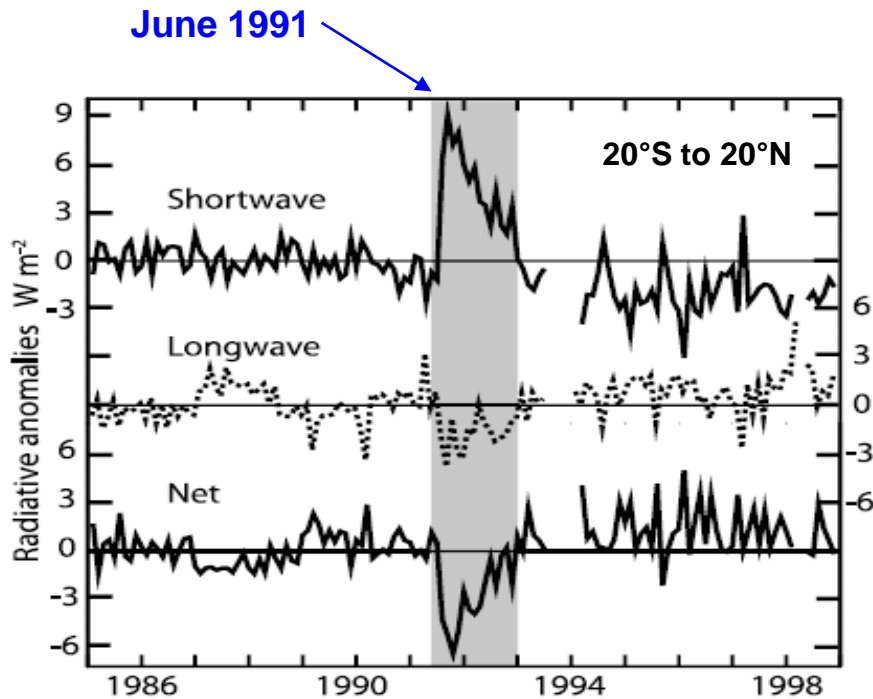
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Trenberth and Dai, *GRL*, 2007

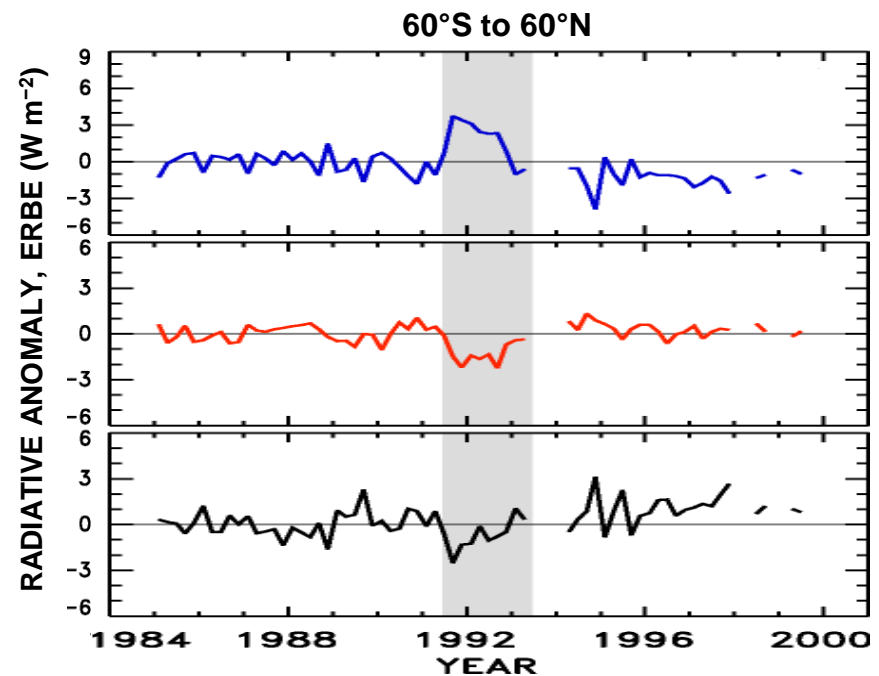
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**Radiative anomaly due to Pinatubo
may have been $\sim 4.5 \text{ W m}^{-2}$
in the tropics**



Trenberth and Dai, *GRL*, 2007

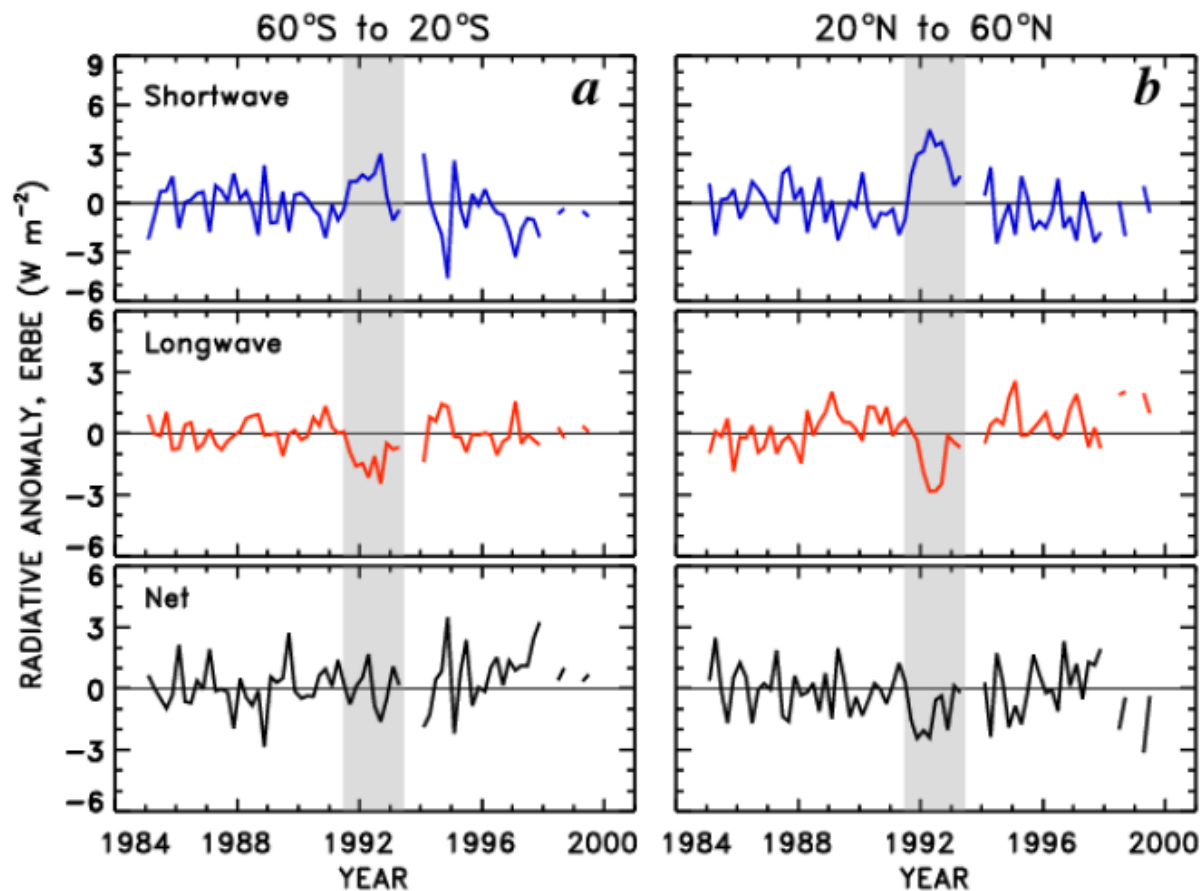
**Global RF anomaly due to Pinatubo
was not as large as $\sim 4.5 \text{ W m}^{-2}$**



Canty et al., *ACP*, 2013

- Mt Pinatubo: $\Delta S_{\text{STRATOSPHERE}} \approx 6 \text{ Tg} \Rightarrow 4.5 \text{ W m}^{-2} \downarrow$ surface radiative forcing
0.5 °C cooling
- Doubling CO₂ will result in $\sim 3.7 \text{ W m}^{-2} \uparrow$ surface radiative forcing

**Almost no net RF anomaly due to Pinatubo
outside of the tropics !**



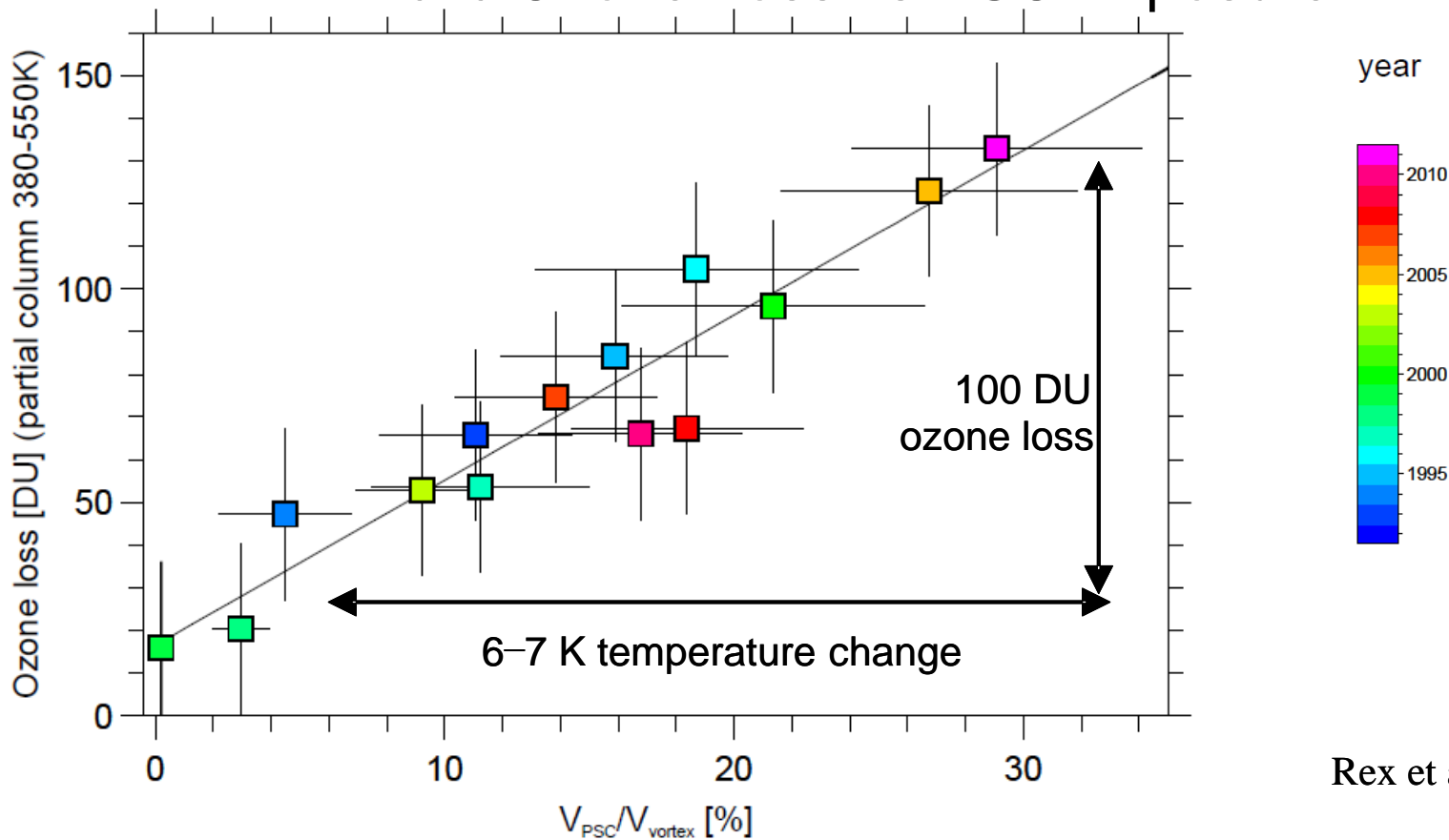
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- Requires **5.3 Tg** perturbation to stratospheric S to counter
 - requires continuous injection of 2.65 to 5.3 Tg S per year (due to 2 or 1 yr $\tau_{\text{STRATOSPHERE}}$)
 - estimated cost \$70 to 140 billion per year (\$70 to 140 per capita of affluent world)
 - for comparison: annual military expenditures \$1000 billion per year
 - advocates manufacture & surface release of a special gas (insoluble, non-toxic, un-reactive with OH, and zero GWP) that is processed photochemically only in the stratosphere to yield sulfate aerosols (he's an atmospheric chemist!)
- Ozone depletion
 - Global column O₃ declined by ~2.5% following eruption of Mt. Pinatubo
 - Compensating for CO₂ doubling would lead to less ozone loss than followed Pinatubo
 - Stratospheric chlorine is declining, so enhanced O₃ loss less worrisome in the future

Will the response of polar ozone to stratospheric sulfur injection be as modest as suggested by the response of global ozone to Mt. Pinatubo aerosol?

Arctic Ozone Loss vs PSC Exposure

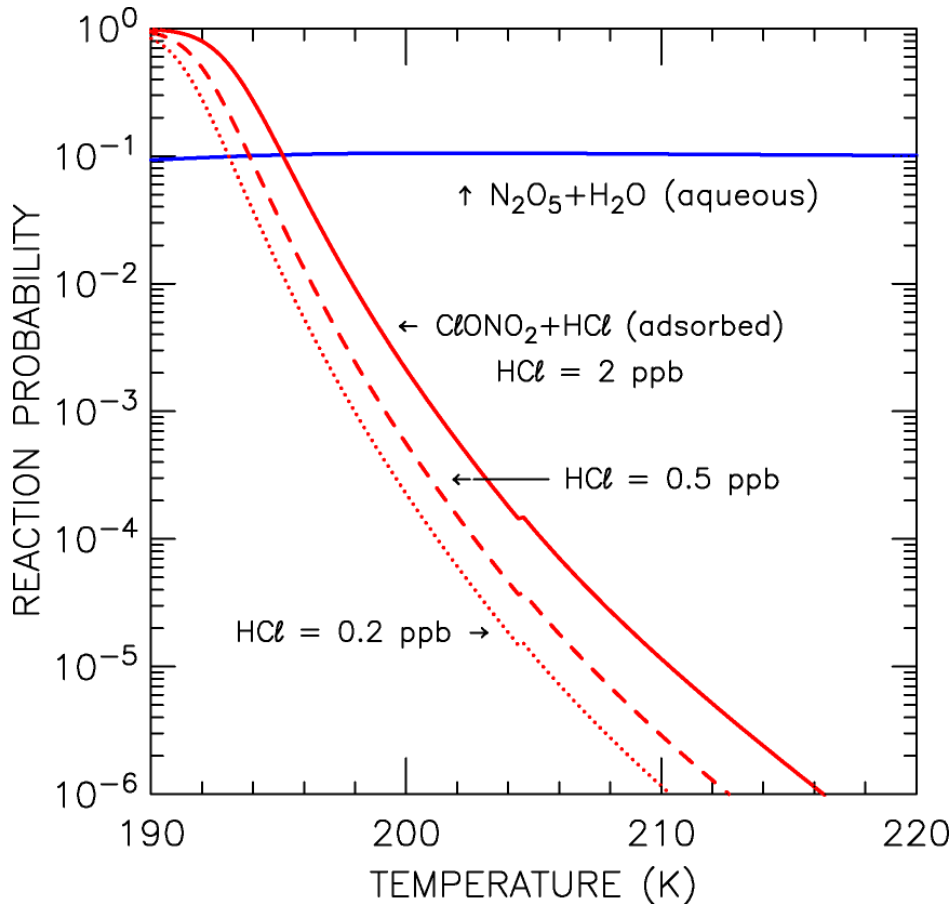


Rex et al., GRL, 2006

PSCs \Rightarrow polar stratospheric clouds: provide **surfaces** for heterogeneous conversion of HCl and ClNO₃ to ClO

Lecture 15

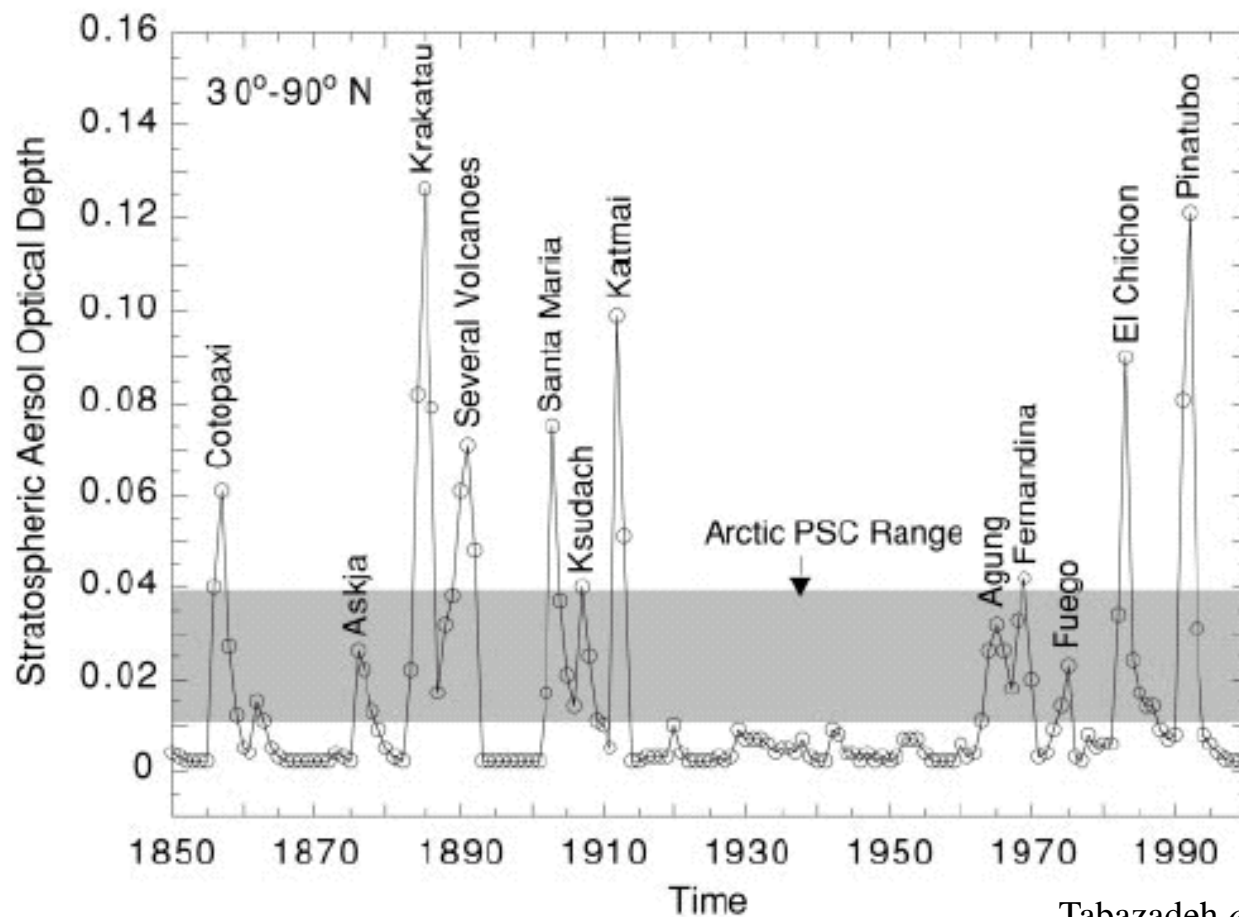
Chlorine Activation



- Chlorine activation reactions occur on cold aerosols
- Chlorine activation depends on T (which drives γ) as well as Surface Area

$$k = \frac{1}{4} \gamma (\text{Velocity}_{\text{ClONO}_2}) (\text{Aerosol Surface Area per Unit Volume})$$

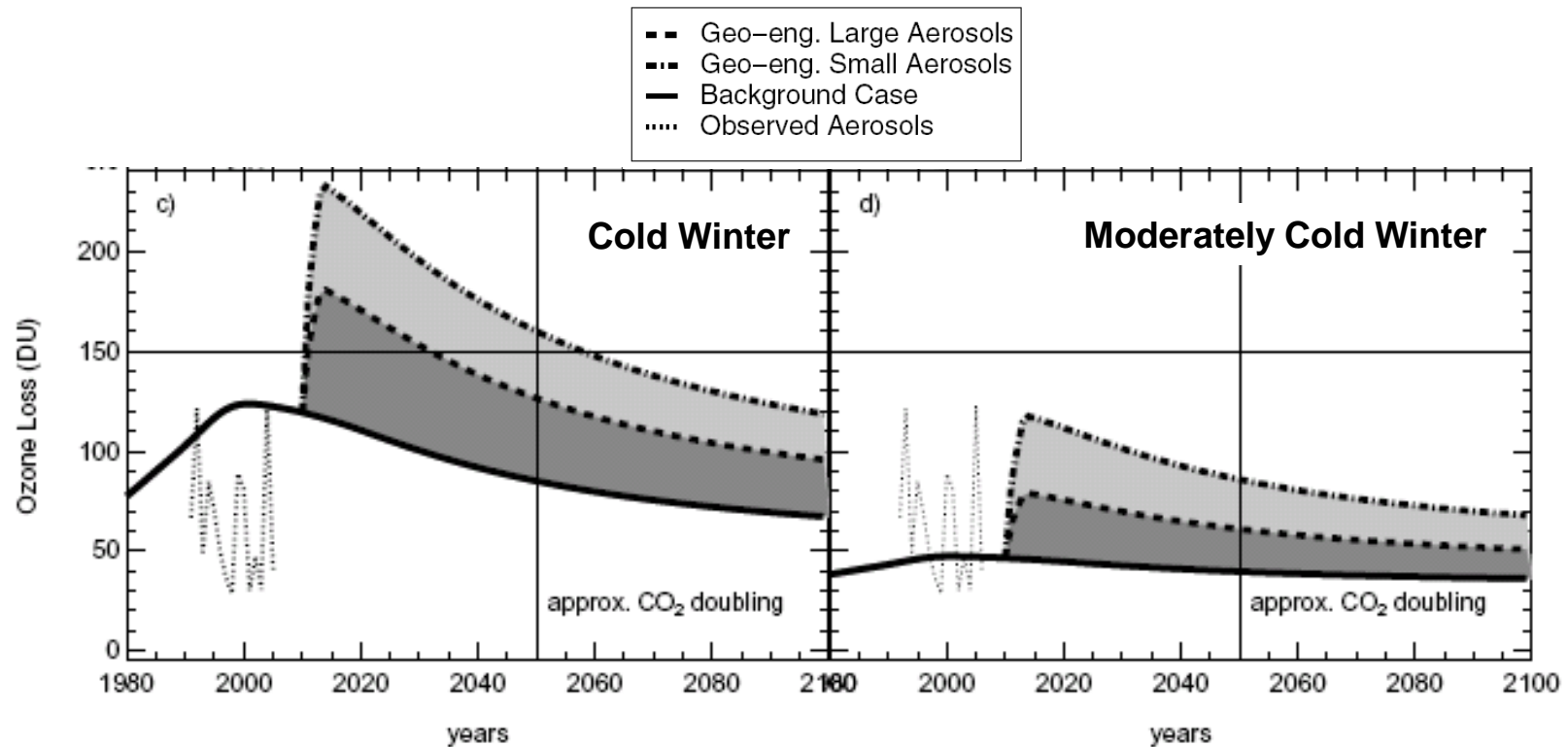
Lecture 11



Tabazadeh *et al.*, PNAS, 99, 2609, 2002

- Chlorine activation reactions occur on cold aerosols
- Chlorine activation depends on T (which drives γ) as well as Surface Area
- Volcanoes provide more reactive surface area than PSCs !

Effect of Geo-Engineering on Arctic O₃ Loss



Enhancement of stratospheric aerosols due to geo-engineering risks:

- a) future *Arctic Ozone Hole* in “cold” winters (i.e., 1995, 1996, 2000, 2005)**
- b) 30 to 70 year delay in the recovery of the Antarctic ozone hole**

Tilmes *et al.*, *Science*, 2008

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- Ozone depletion

- Global column O₃ ↓ 2.5% following eruption of Mt. Pinatubo
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- Stratospheric chlorine is declining, so enhanced O₃ loss less worrisome in the future

- National Academy of Sciences (2009):

For the injection of sulfate aerosols, ***an additional concern exists***: the potential for increased concentrations of stratospheric aerosols to enhance the ability of residual chlorine, left from the legacy of chlorofluorocarbon use, to damage the ozone layer, especially in the early spring months at high latitudes. A sudden increase in stratospheric sulfate aerosol ***could strongly enhance chemical loss of stratospheric polar ozone for several decades, especially in the Arctic*** (Tilmes *et al.*, 2008: 86 citations !)

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▪ National Academy of Sciences (2015):

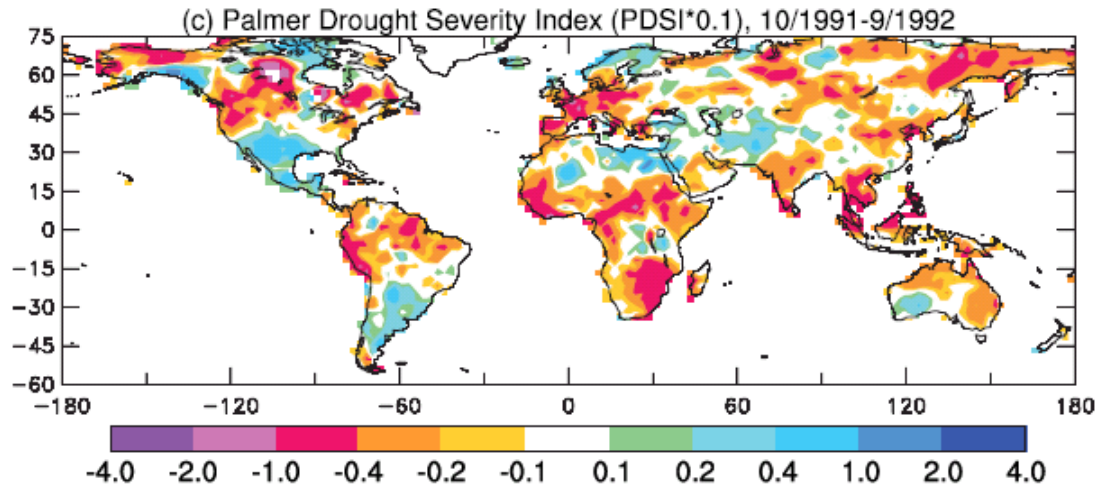
Tilmes et al. (2009; 2008), Heckendorn et al. (2009) and Pitari (2014) explored the impact of SAAM on ozone depletion, and concluded that SAAM (Stratospheric Aerosol Albedo Modification) sufficient to counter a doubling of CO₂ would **delay ozone recovery** (due to the decrease in halogens) by a few decades

Quote from a geo-engineering email thread:

Paul Crutzen's Nobel prize was for his work on the ozone layer; he is in a good position to claim the effect on ozone would not be excessive

Solar Radiation Management: Other Issues

- Enhanced acid precipitation (sulfate will ultimately reach the surface)
- Reducing solar radiation at surface (short wave) may lead to decreased evaporation and precipitation
 - *Precipitation anomalies after Pinatubo suggest risk of widespread drought*



Trenberth and Dai, *GRL*, 2007

Palmer Drought Severity Index for October 1991 to September 1992; warm colors indicate drying. Values less than 0.2 indicate moderate drought, values less than 0.3 indicate severe drought

- **Model calculations (NASA GISS Model E) indicate stratospheric sulfate injections would disrupt the Asian and African summer monsoons, reducing precipitation to area that supply food to billions of people (Robock *et al.*)**
- **If we ever do implement geo-engineering, rapid warming would likely ensue if the perturbation were to stop**

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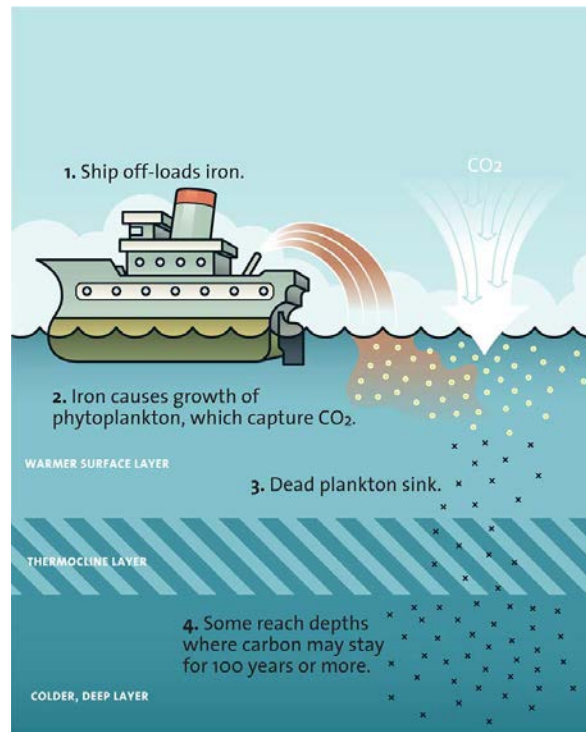
by Paul J. Crutzen : Climatic Change, 77, 211-219, 2006

“Very best if emissions of GHGs could be reduced so that the stratospheric sulfur release experiment would not need to take place. Currently, this looks like a pious wish.”

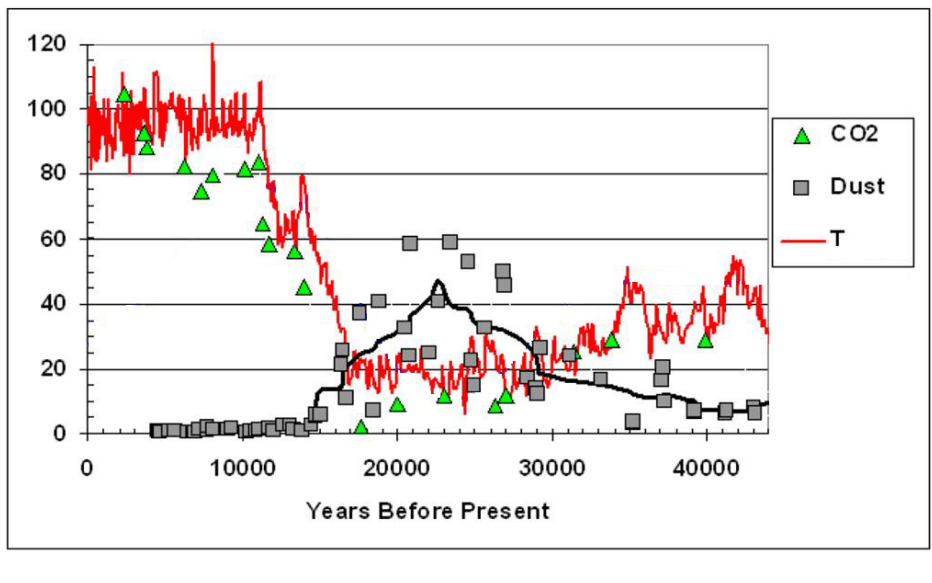
If society is able to successfully “manage solar radiation” reaching the surface, what ecological impact of rising CO₂ would still occur ?

Sequestration of CO₂ from the Atmosphere: Ocean Biology

- Iron's importance to phytoplankton growth and photosynthesis in the ocean dates back to the 1930s, when English biologist Joseph Hart speculated that the ocean's great "desolate zones" (areas apparently rich in nutrients, but lacking in plankton activity or other sea life) might be due to an iron deficiency
- This observation has led to speculation by numerous scientists that "tanker loads" of iron powder, deposited in the right place and time, would increase oceanic dissolved iron content enough to turn these "desolate regions" into oceanic biological havens



<http://www.motherjones.com/files/legacy/news/outfront/2008/03/dumping-iron-1000.jpg>



Vostok ice core data for **changes** in temperature (units of 0.1 K), CO₂ (ppmv), and dust aerosols (linear scale normalized to unity for Holocene) Black line shows 5 point running mean of dust.

Chylek and Lohmann, *GRL*, 2008

Lecture 4

**GLACIAL-INTERGLACIAL CO₂ CHANGE:
THE IRON HYPOTHESIS**

**PALEOCEANOGRAPHY, VOL. 5,
NO. 1, PAGES 1-13 1990**

John H. Martin

In contrast, atmospheric dust Fe supplies were 50 times higher during the last glacial maximum (LGM). Because of this Fe enrichment, phytoplankton growth may have been greatly enhanced, larger amounts of upwelled nutrients may have been used, and the resulting stimulation of new productivity may have contributed to the LGM drawdown of atmospheric CO₂ to levels of less than 200 ppm. Background information and arguments in support of this hypothesis are presented.

Lecture 5

Sequestration of CO₂ from the Atmosphere: Ocean Biology

BOX 3.2

Historical Context of Ocean Iron Fertilization

“Give me half a tanker of iron, and I’ll give you an ice age,” biogeochemist John Martin reportedly quipped in a Dr. Strangelove accent at a conference at Woods Hole in 1988 (Fleming, 2010). Martin and his colleagues at Moss Landing Marine Laboratories proposed that iron was a limiting nutrient in certain ocean waters and that adding it stimulated explosive and widespread phytoplankton growth. They tested their iron deficiency, or “Geritol,” hypothesis in bottles of ocean water, and subsequently experimenters added iron to the ocean in a dozen or so ship-borne “patch” experiments extending over hundreds of square miles (see text for discussion). OIF was shown to be effective at inducing phytoplankton growth, and the question became—was it possible that the blooming and die-off of phytoplankton, fertilized by the iron in natural dust, was the key factor in regulating atmospheric carbon dioxide concentrations during glacial-interglacial cycles? Dust bands in ancient ice cores encouraged this idea, as did the detection of natural plankton blooms by satellites.

This realization led to further questions. Could OIF speed up the biological carbon pump to sequester carbon dioxide? And could it be a solution to climate change? Because of this possibility, Martin’s hypothesis received widespread public attention. What if entrepreneurs or governments could turn patches of ocean green and claim that the carbonaceous carcasses of the dead plankton sinking below the waves constituted biological “sequestration” of undesired atmospheric carbon? Several companies—Climos,¹⁸ Planktos (now out of the business), GreenSea Ventures, and the Ocean Nourishment Corporation¹⁹—have proposed entering the carbon-trading market by dumping either iron or urea into the oceans to stimulate both plankton blooms and ocean fishing (Climos, 2007; Freestone and Rayfuse, 2008; Powell, 2008; Rickels et al., 2012; Schiermeier, 2003).

OIF projects could be undertaken unilaterally and without coordination by an actor out to make a point; in fact, one such incident took place off the coast of Canada in 2012 (Tollefson, 2012). However, as this section describes, there are still unresolved questions with respect to the effectiveness and potential unintended consequences of large-scale ocean iron fertilization.

Sequestration of CO₂ from the Atmosphere: Ocean Biology

- Some scientists have long argued that the iron fertilization vision is flawed because:
 - a) lack of iron not always the limiting factor for growth
 - b) the diatoms that form are much larger than phytoplankton that populate typical surface waters (top of the oceanic food chain)

- Academic research continues:

Biogeosciences, 7, 4017–4035, 2010

Side effects and accounting aspects of hypothetical large-scale Southern Ocean iron fertilization

A. Oschlies¹, W. Koeve¹, W. Rickels², and K. Rehdanz²

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²Kiel Inst. for the World Economy at the Christian-Albrechts Univ. of Kiel, Hindenburgufer 66, 24105, Kiel, Germany

3.7 Ocean acidification

To the extent that OIF sequesters additional CO₂ in the ocean, it will also amplify ocean acidification (Denman, 2008). This is most pronounced in areas where the sequestered CO₂ is stored.

<http://www.biogeosciences.net/7/4017/2010/bg-7-4017-2010.html>

Sequestration of CO₂ from the Atmosphere: Ocean Biology

Maritime Safety

Maritime Security and Piracy

Marine Environment

Pollution Prevention

Pollution Preparedness and Response

Ballast Water Management

Biofouling

Anti-fouling Systems

Ship Recycling

Port Reception Facilities

Special Areas Under MARPOL

Particularly Sensitive Sea Areas

Convention on the Prevention of Marine Pollution by Dumping of Wastes and Other Matter

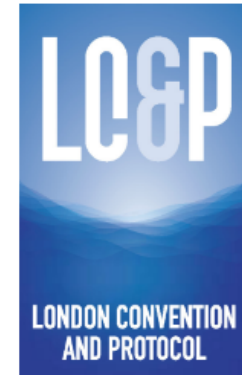
The "Convention on the Prevention of Marine Pollution by Dumping of Wastes and Other Matter 1972", the "London Convention" for short, is one of the first global conventions to protect the marine environment from human activities and has been in force since 1975. Its objective is to promote the effective control of all sources of marine pollution and to take all practicable steps to prevent pollution of the sea by dumping of wastes and other matter. Currently, 87 States are Parties to this Convention.

In 1996, the "London Protocol" was agreed to further modernize the Convention and, eventually, replace it. Under the Protocol all dumping is prohibited, except for possibly acceptable wastes on the so-called "reverse list". The Protocol entered into force on 24 March 2006 and there are currently 48 Parties to the Protocol.

These pages include general information for the public and for States interested in becoming Parties to the London Protocol 1996. Please click on the links to the left for further information on related issues.

Information about the Convention and the Protocol can also be found in the information leaflet (currently available in English only) which contains details on what the London Convention is, achievements to date, the potential benefits and cost of membership, a shortlist of the current activities under the instruments and their relationship with other international agreements.

LC&P



Related Documents

<http://www.imo.org/OurWork/Environment/LCLP/Pages/default.aspx>

Sequestration of CO₂ from the Atmosphere: Ocean Biology



Ocean Fertilization under the LC/LP

Maritime Safety

Maritime Security and
Piracy

Marine Environment

Pollution Prevention

Pollution Preparedness
and Response

Ballast Water
Management

Biofouling

Anti-fouling Systems

Ship Recycling

Port Reception Facilities

Special Areas Under
MARPOL

Particularly Sensitive Sea
Areas

2012

The Scientific Groups met in Jeju, Republic of Korea from 21 to 25 May 2012. A Working Group on Marine Geo-engineering was convened to make recommendations for the development of a web-based repository of references relating to the application of the Assessment Framework for Scientific Research Involving Ocean Fertilization and to develop terms of reference, as appropriate. The Working Group was also tasked to review the feasibility, utility and content of a generic placement assessment, including a revised version submitted by the UK, designed for Contracting Parties to evaluate proposed placement activities. There was no consensus within the Scientific Groups as to the feasibility, utility and content of a generic placement assessment framework or to the benefit of such an approach. However, the groups decided to forward the revised generic placement assessment framework, as set out in annex 2 of LC/SG 35/15, to the LP Intersessional Working Group on Ocean Fertilization and the governing bodies. The UK provided a useful informational paper on the effects of natural iron fertilization on deep-sea ecology.

In considering the work of the Scientific Groups, the governing bodies accepted an offer from the United States to lead the development of a web-based repository of references relating to the application of the Assessment Framework that would be accessible to LC-LP Parties (see LC 34/15). This website is the prototype.

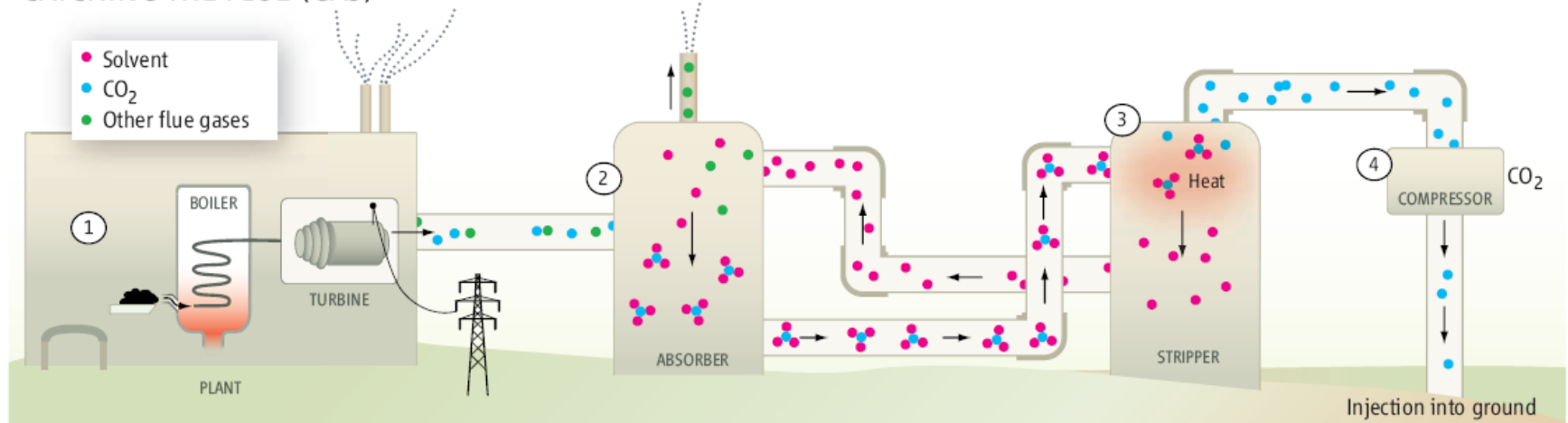
2013

In April 2013, Australia, Nigeria and the Republic of Korea submitted a proposal to amend the London Protocol to regulate placement of matter for ocean fertilization and other marine geo-engineering activities.

The Scientific Groups met in Buenos Aires, Argentina from 27 to 31 May 2013. The Chair of the Correspondence Group reported on progress made towards developing a web-based repository of references relating to the application of the Assessment Framework. A working group was convened to provide feedback to the Correspondence Group and proposed that the team continue its work with a view to having a website ready to share with the governing bodies at their next joint session in October 2013. The Scientific Groups endorsed this proposal and accepted an offer from the United States to re-establish the Correspondence Group (see LC/SG 36/16, para. 3.8). The United Kingdom provided a useful review of the effectiveness, environmental impacts and emerging governance of ocean fertilization, and the United States shared state of the science fact sheet on climate engineering prepared by its National Oceanic and Atmospheric Administration.

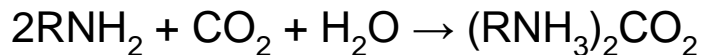
Sequestration of CO₂ from Power Plants

CATCHING THE FLUE (GAS)



How a retrofit works. (1) Most coal plants burn coal to create steam, running a turbine that produces electricity. After treatment for pollutants, the flue gas, a mixture of CO₂ (blue) and other emissions (green), goes out a smokestack. To collect CO₂ for storage, however, the mixture of gases is directed to an absorber (2), where a solvent like MEA (pink) bonds with the CO₂ molecules. The bonded CO₂-solvent complexes are separated in the stripper (3), which requires heat. More energy is needed for the next step (4), which produces a purified CO₂ stream for ground storage as well as solvent molecules that can be reused. (Schematic not to scale.)

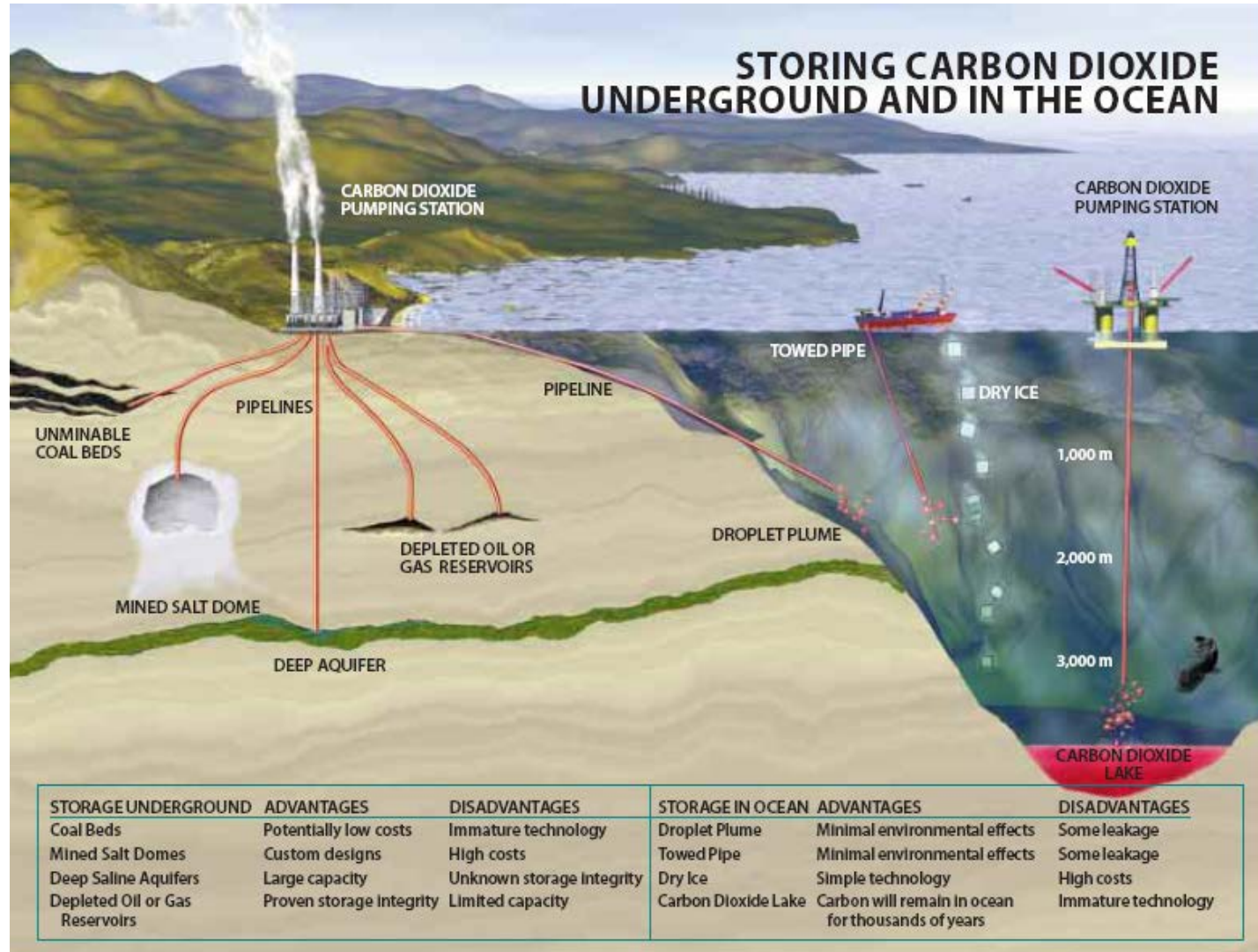
MEA-monoethanolamine (CH₂CH₂OH)NH₂ in an aqueous solution will absorb CO₂ to form ethanolammonium carbamate.



MEA is a weak base so it will re-release the CO₂ when heated

Kintisch, *Science*, 2007

Sequestration of CO₂ from Power Plants



STORAGE SITES for carbon dioxide in the ground and deep sea should help keep the greenhouse gas out of the atmosphere where it

now contributes to climate change. The various options must be scrutinized for cost, safety and potential environmental effects.

Herzog *et al.*, *Scientific American*, 2000

Sequestration of CO₂ from Power Plants

Sleipner, Norway



National Geographic, June 2008

- North Sea natural gas field: enormous capacity
- Captures ~90% of CO₂ that is generated
- CO₂ pumped into 200 m thick sandstone layer 720 m below sea floor
- Project initiated in response to \$50 ton tax on CO₂ emissions instituted by Norwegian Government in 1996
- Investment in capital cost paid off in about one and a half years !

Sequestration of CO₂ from Power Plants: Cost

CCS component	Cost range	
Capture from a power plant	15–75 US\$/tCO ₂ net captured	~\$45/ ton
Capture from gas processing or ammonia production	5–55 US\$/tCO ₂ net captured	
Capture from other industrial sources	25–115 US\$/tCO ₂ net captured	
Transportation	1–8 US\$/tCO ₂ transported per 250km	~\$4.5/ ton
Geological storage	0.5–8 US\$/tCO ₂ injected	~\$4.5/ ton
Ocean storage	5–30 US\$/tCO ₂ injected	
Mineral carbonation	50–100 US\$/tCO ₂ net mineralized	

Cost of capture: **~\$54 / ton CO₂** × 10 × 10⁹ tons C / yr = **\$ 540 billion**

Present cost of fossil fuel: \$ 46 / barrel ≈ **\$ 400 / ton**

World GDP, 2014: **\$ 77.8 trillion** CO₂ capture = 0.7 % of world GDP
or 14 % of cost, barrel of oil

↑
Back of the envelope analysis
←



INTERGOVERNMENTAL PANEL ON CLIMATE CHANGE (IPCC)

Carbon Dioxide Capture and Storage



<http://www.ipcc.ch/pdf/presentations/briefing-montreal-2005-11/presentation-special-report-co2.ppt>

Afforestation

- If 100,000 km² (size of Ireland) was re-planted every year, for 40 years (size of Australia) would sequester between 20 and 50 Gt of C from the atmosphere
- ⇒ between **5** and **10** % of emissions, 2015 to 2055
- Land available ✓ Cost ✓
- But Houghton cautions:
 - forests are dark ... as albedo declines, T rises, particularly in winter
 - once trees are fully grown, sequestration stops (yikes)
 - offset is small fraction of total projected C emission and we have used an area the size of Australia (yikes yikes)



<http://www.worldlandtrust.org/images/places/brazil/wetland-before-after-joy-and-mick-braker-vl.jpg>

Sequestration of CO₂ from the Atmosphere: Burial of Trees

- Prof Ning Zeng (UMCP) advocates planting, harvesting, and burial of rapidly growing trees (proposal is to collect dead trees on forest floor and selectively log live trees)
- Meetings have been held to discuss this idea:
- A UMd Gemstone Project has addressed this issue
<http://teams.gemstone.umd.edu/classof2010/carbonsinks>



- Statements from Zeng, Carbon Sequestration Via Wood Burial, *Carbon Balance and Management*, 2008 <http://www.cbmjournals.com/content/3/1/1> :
 - Here I suggest an approach in which wood from old or dead trees in the world's forests is harvested & buried in trenches under a layer of soil, where the anaerobic condition slows the decomposition of the buried wood.
 - Because of low oxygen below the soil surface, decomposition of buried wood is expected to be slow

National Academy of Sciences (2015) Summary Table

TABLE S.1 Overview of general differences between Carbon Dioxide Removal (CDR) proposals and Albedo Modification proposals. GHG stands for greenhouse gases released by human activities and natural processes and include carbon dioxide, methane, nitrous oxide, chlorofluorocarbons and others. The Committee intends to limit discussion to proposals that raise the fewest problematic issues, thus excluding ocean iron fertilization from the CDR list. Each statement may not be true of some proposals within each category.

Carbon Dioxide Removal proposals...	Albedo Modification proposals...
... address the cause of human-induced climate change (high atmospheric GHG concentrations).	...do not address cause of human-induced climate change (high atmospheric GHG concentrations).
...do not introduce novel global risks.	... introduce novel global risks.
...are currently expensive (or comparable to the cost of emission reduction).	...are inexpensive to deploy (relative to cost of emissions reduction).
...may produce only modest climate effects within decades.	...can produce substantial climate effects within years.
...raise fewer and less difficult issues with respect to global governance.	...raise difficult issues with respect to global governance.
...will be judged largely on questions related to cost.	...will be judged largely on questions related to risk.
...may be implemented incrementally with limited effects as society becomes more serious about reducing GHG concentrations or slowing their growth.	...could be implemented suddenly, with large-scale impacts before enough research is available to understand their risks relative to inaction.
...require cooperation by major carbon emitters to have a significant effect.	...could be done unilaterally.
...for likely future emissions scenarios, abrupt termination would have limited consequences.	...for likely future emissions scenarios, abrupt termination would produce significant consequences.